Randomization of chemical reaction networks

Christoph Flamm

xtof@tbi.univie.ac.at

Institute for Theoretical Chemistry University of Vienna

> joined work with: Philipp Honegger (HMS) Walter Fontana (HMS)

Bled, February 16 2023



Widespread use of random chemistry models

SCIENCE ADVANCES | RESEARCH ARTICLE

SYSTEMS BIOLOGY

Universal scaling across biochemical networks on Earth

Hyunju Kim^{1,2}*, Harrison B. Smith²*, Cole Mathis^{1,3}, Jason Raymond², Sara I. Walker^{1,2,4,5†}

Kim et al 2019, Sci Adv 5:eaau0149 | doi:10.1126/sciadv.aau0149

Scaling laws in enzyme function reveal a new kind of biochemical universality

Dylan C. Gagler^a, Bradley Karas^a(), Christopher P. Kempes^b, John Malloy^a(), Veronica Mierzejewski^a(), Aaron D. Goldman^c(), Hyunju Kim^{a,d,e,1}(), and Sara I. Walker^{a,b,d,e,1}()

Gagler et al 2022, PNAS 119(9):e2106655119 | doi:10.1073/pnas.2106655119



J. R. Soc. Interface (2012) 9, 1168-1176 doi:10.1098/rsif.2011.0652 Published online 30 November 2011

Evolutionary significance of metabolic network properties

G eorg B asler^{1,*}, Sergio G rimbs¹, Oliver E benhöh², J oachim Selbig^{1,3} and Zoran N ikoloski^{1,3}

Mathematical random network models

Purpose of prototype models:

- null models (statistic significance of observed features).
- insight how observed features arise from construction rules.

The most common prototype models:

- 1 Erdös-Rényi (ER) Model (small-world).
- 2 Watts-Strogatz (WS) Model (small world + local clustering).
- 3 Barabási-Alberts (BA) Model (scale-free).

Basic assumption: A nodes can interact with any other node.

Assumption fails for chemistry!!

Albert R & Barabási A-L (2002), Statistical mechanics of complex networks, Rev Mod Phys 74:47-97

Randomization strategies for chemical networks

1 Randomize network structure

- a. construct a large network instance
- b. thin out network by random
 - vertex sampling.
 - edge sampling.
 - walk sampling.

2 Randomize network chemistry

- reaction preception method.
- mechanistic constraints on rections.

Q1: Are networks generated by these strategies chemical? A: Not necessarily, only if all reactions are mass balanced =; (

Q2: Is every directed hypergraph a chemical network? A: No, only if the network is conservative[†] (i.e. mass preserving)!

 $\exists m \gg 0$ sucht that $m^T \cdot S = 0$

[†] Stefan Müller, Christoph Flamm, Peter F Stadler, What makes a reaction network "chemical"? J Cheminfo 14:63, 2022 doi:10.1186/s13321-022-00621-8

What to preserve in chemical reactions

... apart from mass?



Rewiring strategies 1/2



Ор	MF	Mdeg	Rdeg	Mdeg _{tot}
111	Y	Y	Y	Y
110	Y	Y	N	Y
101	Y	N	Y	Y
100	Y	N	N	Y

Operation 3-digit code:

1st digit ... one-sided (only educt or product side of reaction)

2nd digit ... role change (educts become products and vice versa)

3rd digit ... arity (number of educts and products of a reaction)

 $Op\ldots$ operation; $MF\ldots$ molecular formula; $Mdeg\ldots$ in/out degree of molecule node; $Rdeg\ldots$ reaction node degree; $Mdeg_{tot}\ldots$ total degree of molecular node

Rewiring strategies 2/2



010 - Two-sided, no role-change, vary arity







000 - Two-sided, allow role-change, vary arity



Ор	MF	Mdeg	Rdeg	Mdeg _{tot}
111	Y	Y	Y	Y
110	Y	Y	N	Y
101	Y	N	Y	Y
100	Y	N	N	Y

Operation 3-digit code:

1st digit ... one-sided (only educt or product side of reaction)

2nd digit ... role change (educts become products and vice versa)

3rd digit ...arity (number of educts and products of a reaction)

 $Op\ldots$ operation; $MF\ldots$ molecular formula; $Mdeg\ldots$ in/out degree of molecule node; $Rdeg\ldots$ reaction node degree; $Mdeg_{tot}\ldots$ total degree of molecular node

(Auto)Catalyst as Sets $x + y + z \xrightarrow{\{A,B,C\}} x' + y' + z'$

- Rate acceleration: r₁ r₃ must together proceed faster than the spontanious process.
- 2 Set of catalysts is conserved.
- **3** Each reaction involve:
 - catalysts.
 - at least 1 catalyst as reactant.
 - at least 1 catalyst as product.
- The production of a species from the set of catalysts depends on the presence of another species from the set of catalysts.



0+0 induce the presence of a cycle. Species 0 show turnover, species 0 remain conserved. For catalysis ignore gray dashed arrows.

The stoichiometric matrix S



Note that a catalyst, which enters and exits a reaction with the same stoichiometry, has as well a zero entry in S. The yellow highlighted region is a restriction of the S to the autocatalytic cycle. The species in the cyan region are considered externl to the autocatalytic cycle, and are thought to be chemo-stated.

Primer: Carbohydrate Chemistry $(CH_2O)_n$ Sugars are organic compounds with an C:H:O ratio of 1:2:1.

The reactivity of sugars is largely dominated by the carbonyl group (C=O) and the vicinal alcohol groups (HO-C-C-OH).

The keto-enol isomerization reaction and the aldol condensation a C-C bond formation reaction are of importance.





Type 1 autocatalytic core: A keto-enole isomerization, **A** Aldol / retro-Aldol reaction. Note that the reaction sequence from glycolealdehyde to erytrose is compressed into a single reaction (r_1) in the type 1 autocatalytic core figure. Butlerov AM (1861), Eninges über die chemische Structur der Körper, Zeitschrift für Chemie 4:549-560;

Amino acid thioesters, cystamine autocatalytic system



Type 3 autocatalytic core; The reaction chemistry is thiol-disulfide exchange □, thiol-thoester exchange □ and native chemical ligation □; orange arrows connect food or wast molecules to the autocatalytic cycle, which has a type 3 topology; cystamine (in the center) is the "autocatalyst"; [Semenov et al Nature 2016]

Rebeck's Autocatalytic System



Type 3 autocatalytic core; Molecules in cyan boxes are sequestered in a molecular cage. The synthesis of the autocatalytic species and its storage, occures temporally, spatially, and chemically distinct from the process that depletes the store in an autocatalytic fashion. [Chen et al PNAS 2001]



Finding Autocatalysis in reaction networks $_{\text{B}}$





- A Find autocatalytic skeleton:
 - 1 Find fission and merging points.
 - 2 Connect fission and merginf points by paths.
- B Embed autocatalytic skeleton in hypergraph:
 - 1 Insert induced edges (highlighted in green labeled i).
 - 2 Add species and edges attached to sceleton reaction nodes. (highlighted in yellow; nodes: ii immediate substrates and products; iii mediator species).

Poly-unsaturated fatty acid system (PUFA)



Red bull's eye marks autocatalytic species. Reactions are yellow boxes. Skeleton connected by black arrows. Red / green cross-links are introduced in embeding phase and render the network catalytic.

Using Strategy 111 to randomize GEMs



Chlamydomonas reinhardtii

Single-celled green alga. Extended lipid metabolism.



Conclusion

- A formalism for Chemistry must be expressive enough to capture the major characteristics of reactive systems:
 - mass conservation.
 - structural change.
 - atom-to-atom mappings.
- Q Graph grammar formalization has the right level of abstraction.
- 3 Chemical reaction space is vast, hence:
 - computational exploration is indispensable.
 - hyperflows allows to search for complex reaction patterns.
 - automated atom tracking in reaction networks is important.
- Gomputational methods can be a powerful way to gain insight into complex chemistry.

Mathematical Modelling for Microbial Community Induced Metabolic Diseases



!!! Postdoc Position availabel in Vienna !!!

Novo Nordisk Foundation grant NNF21OC0066551 (45 mio. DKK, 2022-2028).

Further reading



Andersen JL, Flamm C, Merkle D, Stadler PF.

In silico Support for Eschenmoser's Glyoxylate Scenario. Isr J Chem 55:919-933 (2015) | doi:10.1002/ijch.201400187



Andersen JL, Flamm C, Merkle D, Stadler PF.

Defining Autocatalysis in Chemical Reaction Networks. J Sys Chem 8:121-133 (2020) | Preprint BIOINF 20-004



Barenholz U, Davidi D, Reznik E, Bar-On Y, Antonovsky N, Noor E, Milo R.

Design principles of autocatalytic cycles constrain enzyme kinetics and force low substrate saturation at flux branch points.

eLife 6:e20667 (2017) | doi:10.7554/eLife.20667



Bissette AJ, Fletcher SP.

Mechanisms of Autocatalysis. Angew Chem Int Ed 52:12800-12826 (2013) | doi:10.1002/anie.201303822



Blokhuis A, Lacoste D, Nghe P.

Universal motifs and the deversity of autocatalytic systems. PNAS 117(41):25230-25236 (2020) | doi:10.1073/pnas.2013527117



Chen J, Körner S, Craig SL, Lin S, Rudkevich DM, Julius Rebek J Jr. Chemical amplification with encapsulated reagents. PNAS 99(5):2593-2596 2001 | doi:10.1073/pnas.052706499



Hanopolskyi AI, Smaliak VA, Novichkov AI, Semenov SN. Autocatalysis: Kinetics, Mechanisms and Design. ChemSystemsChem 2:e2000026 (2020) | doi:10.1002/syst.202000026



Semenov SN, Kraft LJ, Ainla A Zhao M Baghbanzadeh M, Campbell VE, Kang K, Fox JM, Whitesides GM. Autocatalytic, bistable, oscillatory networks of biologically relevant organic reactions. Nature 537:656-660 2016 | doi:10.1038/nature19776