Leontis Westhof Notation for RNA-Protein Complexes?

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27th TBI Winterseminar in Bled

Motivation		Methods	Results	Outlook
Obiective	and Motivation	I		

- RNA molecules present a vital part of life
- Secondary structure vs. tertiary interactions
- Growing number of 3D RNA structures enables systematic investigation of tertiary interactions
- Led to catalogs of non-WC base pairs and definition of recurring 3D modules [1, 2, 3]
- How do nucleotides interact with amino acids?

Motivation	Background	Methods	Results	Outlook
	 <u>.</u>			

Classification of base pairs

Base pair classification by Saenger in '84 [4]



taken from http://ndbserver.rutgers.edu/atlas/legends/saenger.html

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 Motivation
 Background
 Methods
 Results
 Outlook

 Classification of base pairs (cont.)

Leontis Westhof (LW) notation defines 3 edges [2]



• Other bp classifications exists like in Lee and Gutell, 2004 [3]

Motivation	Background	Methods	Results	Outlook
Hydrogen	bonding			

- Base pairs are established via hydrogen bonding (H-bonds)
- H-bonds contribute to stability of 3D fold
- Electronegative atoms form donor/acceptor pairs

Motivation			Ba	ackgro	und				lethod				Resi	ılts			Οι	
Hydrog	gen	bo	ond	ing														
Period (horizontal)																		
1	Н 2.20																	He
2	Li 0.98	Be 1.57											В 2.04	C 2.55	N 3.04	0 3.44	F 3.98	Ne
3	Na 0.93	Mg 1.31											Al 1.61	Si 1.90	Р 2.19	S 2.58	Cl 3.16	Ar
4	K 0.82	Ca 1.00	Sc 1.36	Ті 1.54	V 1.63	Cr 1.66	Mn 1.55	Fe 1.83	Co 1.88	Ni 1.91	Cu 1.90	Zn 1.65	Ga 1.81	Ge 2.01	As 2.18	Se 2.55	Br 2.96	Kr 3.00
5	Rb 0.82	Sr 0.95	Y 1.22	Zr 1.33	Nb 1.6	Mo 2.16	Тс 1.9	Ru 2.2	Rh 2.28	Pd 2.20	Ag 1.93	Cd 1.69	In 1.78	Sn 1.96	Sb 2.05	Те 2.1	I 2.66	Xe 2.60
6	Cs 0.79	Ba 0.89	*	Hf 1.3	Та 1.5	W 2.36	Re 1.9	Os 2.2	Ir 2.20	Pt 2.28	Au 2.54	Hg 2.00	Tl 1.62	Pb 2.33	Bi 2.02	Po 2.0	At 2.2	Rn 2.2
7	Fr 0.7	Ra 0.9	**	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	Uuo

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H-atoms missing in most 3D structures

Motivation	Background	Methods	Results	Outlook
Dataset				
∎ 3D	Structures from Pro	otein Data Bank		
	Polymer Ty	ype		





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Motivation	Background	Methods	Results	Outlook
Dataset				
■ 3D Struc	ctures from Proteir	n Data Bank		

Polymer Type



60.4% Protein/DNA (2181 hits) 32.8% Protein/RNA (1184 hits) 2.8% Protein/DNA/RNA (101 hits) 2.8% DNA-RNA Hybrid (100 hits) 1.2% DNA/RNA (43 hits)

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Predicting non-canonical base pairs

Computational identification of base pairs

- Reference structures give empirical values for
- 1 distances between (donor/acceptor) atoms
- 2 angles between planes of bases

Motivation	Background	Methods	Results	Outlook

Characterizing the geometry of nucleic acids interacting with proteins, obviously, brings up a whole new host of geometrical issues.

Olson et al., 2001 [5]

Motivation		Methods	Results	Outlook
Predicting	H-bonds			

Scanning for putative H-bonds

- Donor/acceptor atoms: N, O, [C, S, Se]
- Minimum euklidean distance to find putative atom pair
- Distance cutoff ($d \leq 3$ angstrom)
- Penalty for atoms from adjacent nucleotides

	Methods	Results	Outlook
anda			
	Background	Background Methods	Background Methods Results

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Recap: H atoms missing

Motivation	Methods	Results	Outlook

H-bond donor/acceptor atoms



Motivation	Methods	Results	Outlook

H-bond donor/acceptor atoms



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Motivation		Methods	Results	Outlook
Effect of miss	sing H atoms			

- Use all structures that contain H atoms
- Treat them as:
 - a) containing no H atoms
 - b) containing H atoms
- Compare predicted H-bonds

Motivation		Methods	Results	Outlook
Effect of	missing H atom	S		
Effect of	missing H atom	S		

- Use all structures that contain H atoms
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 - a) containing no H atoms
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Calculate number of equal and different predictions b

X-H type: add 1 to $\begin{cases} equal & \text{if } b(X) = b(H) \\ different & \text{if } b(X) \neq b(H) \end{cases}$

Motivation		Methods	Results	Outlook
Effect of mis	ssing H atoms			

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X-H type: add 1 to
$$\begin{cases} equal & \text{if } b(X) = b(H) \\ different & \text{if } b(X) \neq b(H) \end{cases}$$

H₁-**X-H**₂ type: add 1 to
$$\begin{cases} equal & \text{if } b(X) = b(H_i) \\ different & \text{if } b(X) \neq b(H_i) \\ different & \text{if } b(X) \neq b(H_i) \end{cases}$$

Motivation		Methods	Results	Outlook
Quality of H k	oond prediction			

Number of equal and different predictions

type	all	equal	%	different	%
X-H	4021	3499	0.87	522	0.13
H_1 -X- H_2	2758	1600	0.58	1158	0.42

Motivation		Methods	Results	Outlook
H-bond dist	ribution			

Relative frequency of having an H-bond with another nucleotide or amino acid

)2
13
J4/O4
5
)2'
)5'
)
)P1
)P2



Nucleotide-Nucleotide

Nucleotide-Amino Acid



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Motivation	Methods	Results	Outlook
Outlook			

- More sophisticated model for H-bonds
- Redundancy/Bias of structures in PDB
- Definitions of pair and stacking interactions
- Empirical rules for nucleotide amino acid interactions
- Do we see all possible interactions in the PDB data?

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References I			

- N.B. Leontis and E. Westhof.
 - Analysis of rna motifs.

Current opinion in structural biology, 13(3):300–308, 2003.

N.B. Leontis and E. Westhof.

Geometric nomenclature and classification of rna base pairs. *Rna*, 7(4):499–512, 2001.

J.C. Lee and R.R. Gutell.

Diversity of base-pair conformations and their occurrence in rrna structure and rna structural motifs.

Journal of molecular biology, 344(5):1225–1249, 2004.

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References II				
📄 W. Saer	iger.			

- vv. Jaenger.
 - Principles of nucleic acid structure. Springer-Verlag New York, 1984.
- W.K. Olson, M. Bansal, S.K. Burley, R.E. Dickerson, M. Gerstein, S.C. Harvey, U. Heinemann, X.J. Lu, S. Neidle, Z. Shakked, et al.

A standard reference frame for the description of nucleic acid base-pair geometry.

Journal of molecular biology, 313(1):229–237, 2001.