

Dynamische Strukturbildung und Systemrisiko in biologischen Systeme

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Web-Page für weitere Informationen:

<http://www.tbi.univie.ac.at/~pks>

Peter Schuster. 2015. Ebola – Challenge and Revival of Theoretical Epidemiology.
Complexity **20**(5):7-12

Peter Schuster. 2016. Major Transitions in Evolution and Technology.
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Peter Schuster. 2016. Some Mechanistic Requirements for Major Transitions.
Phil.Trans.R.Soc.B **371**(1701):e20150439

Peter Schuster. 2016. Increase in Complexity and Information through Molecular Evolution. *Entropy* **18**(11):e397

Autocatalysis is rare in chemistry but obligatory in biology.

Biological organisms store information in encoded form and have a record of their history.

Particle numbers are large in chemistry and small in biology.

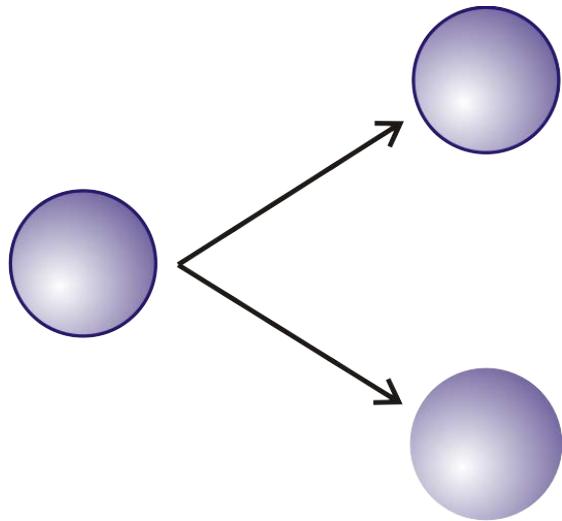
Stochastic phenomena have relatively little importance in chemistry but dominate biology.

What distinguishes biology from chemistry?

Sources of risk relevant uncertainties in predictions:

- (i) exponential growth,
- (ii) complex internal dynamics,
- (iii) multiple quasistationary states,
- (iv) reintroduction of extinguished species.

Autocatalysis, exponential growth, and prediction



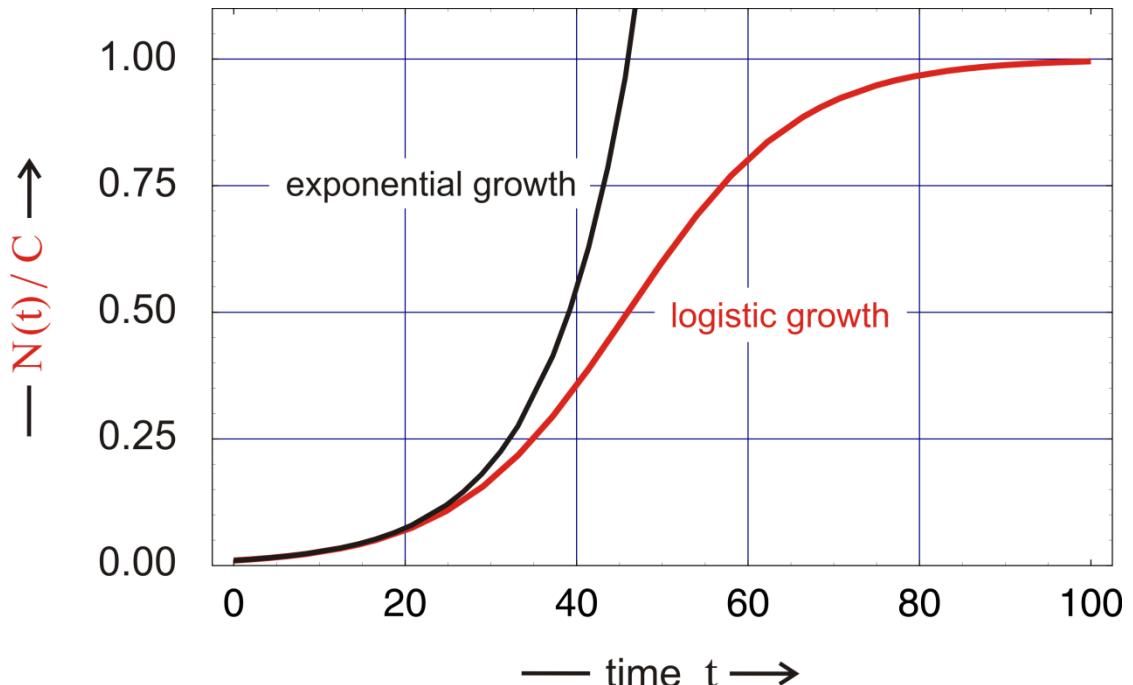
$$(A) + X = 2X$$

$$\frac{dN}{dt} = f x \Rightarrow N(t) = N(0) e^{f t}$$

exponential growth

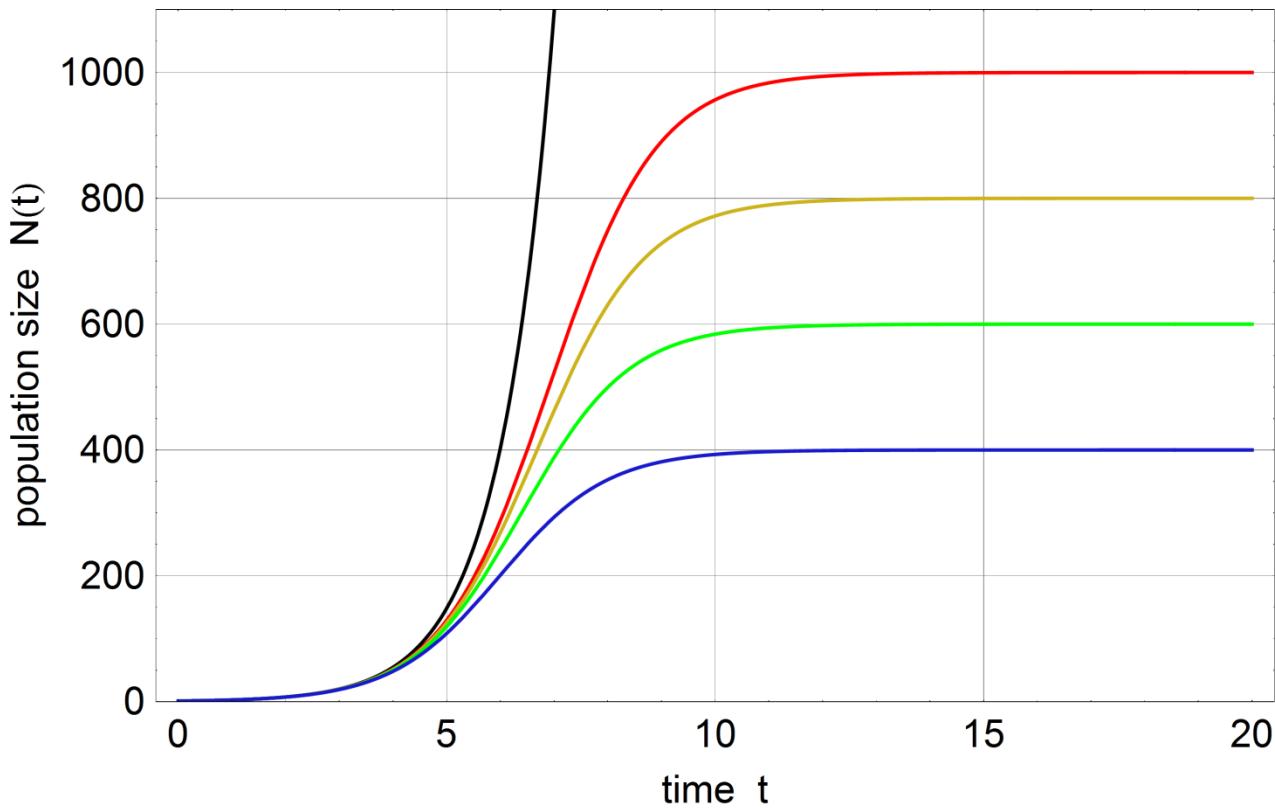


Pierre-François Verhulst,
1804 - 1849



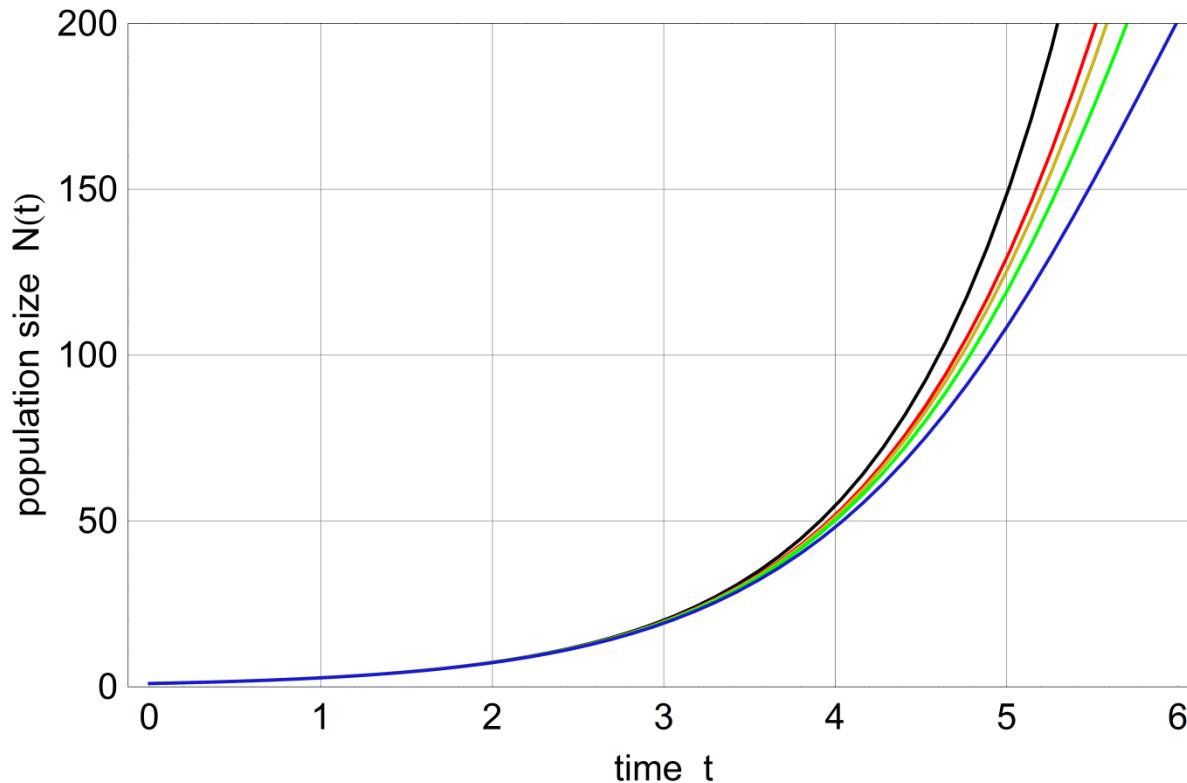
$$\frac{dN}{dt} = f N \left(1 - \frac{N}{C}\right) \Rightarrow N(t) = N(0) \frac{C}{N(0) + (C - N(0)) e^{-f t}}$$

The logistic equation has been conceived in 1838.



Logistic growth with different carrying capacity:

$$C = \infty, 1000, 800, 600, 400$$

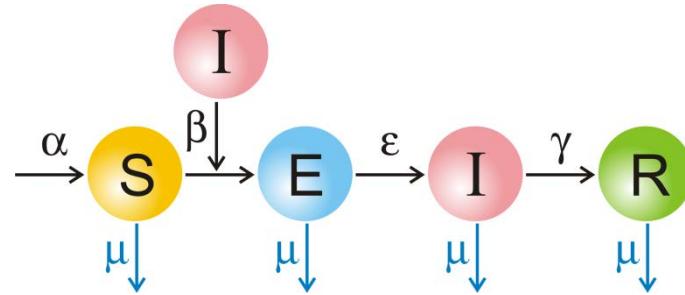


Enlargement of the logistic curves:

$$C = \infty, 1000, 800, 600, 400$$

A.A. King, M.D. de Cellès, F.M.G. Magpantay, P. Rohani. Avoidable errors in the modelling of outbreaks of emerging pathogens, with special reference to Ebola. Proc.Roy.Soc.B 282:e20150347

Complex dynamics and prediction



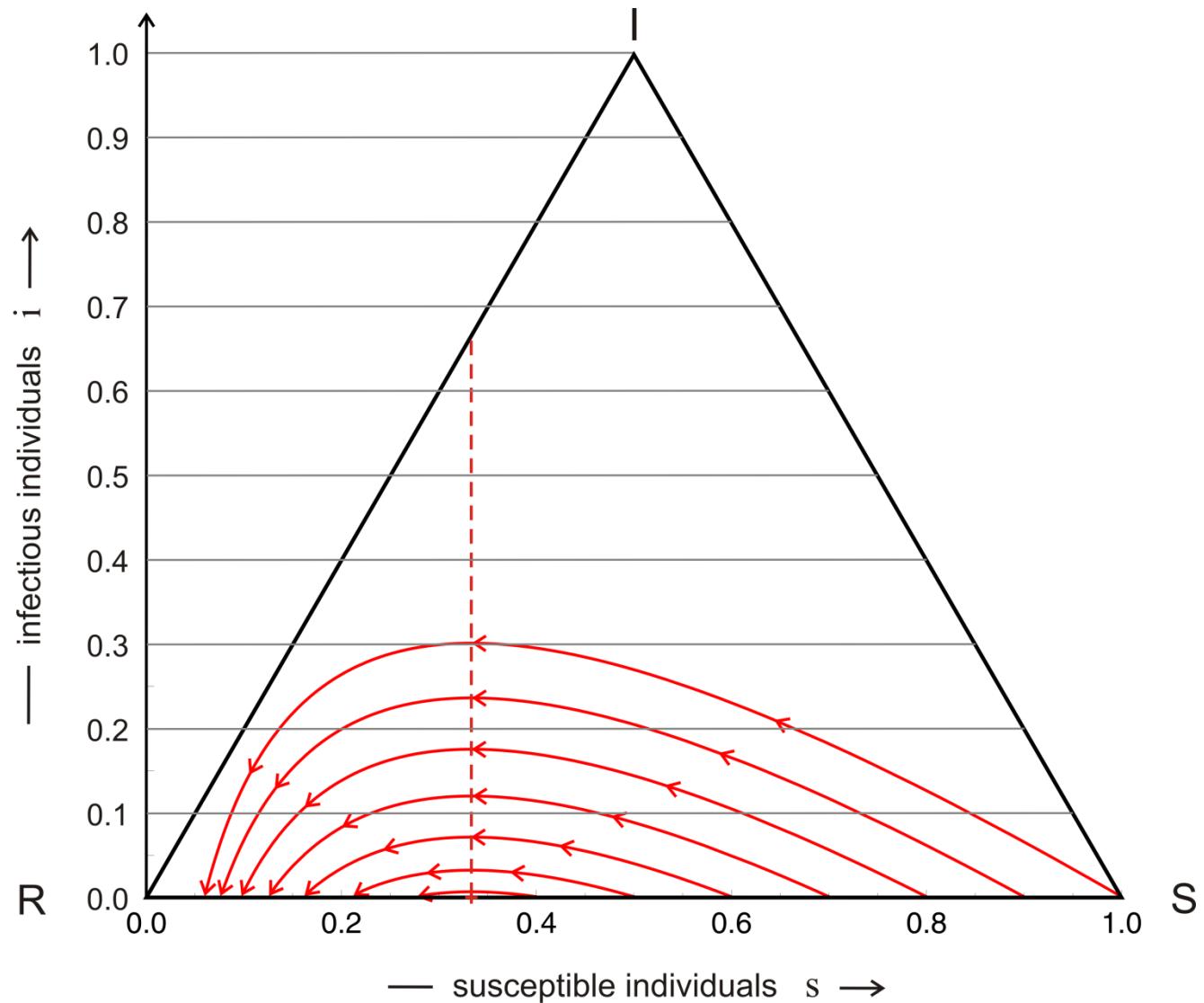
$$\frac{dS}{dt} = \alpha - \beta S I - \mu S$$

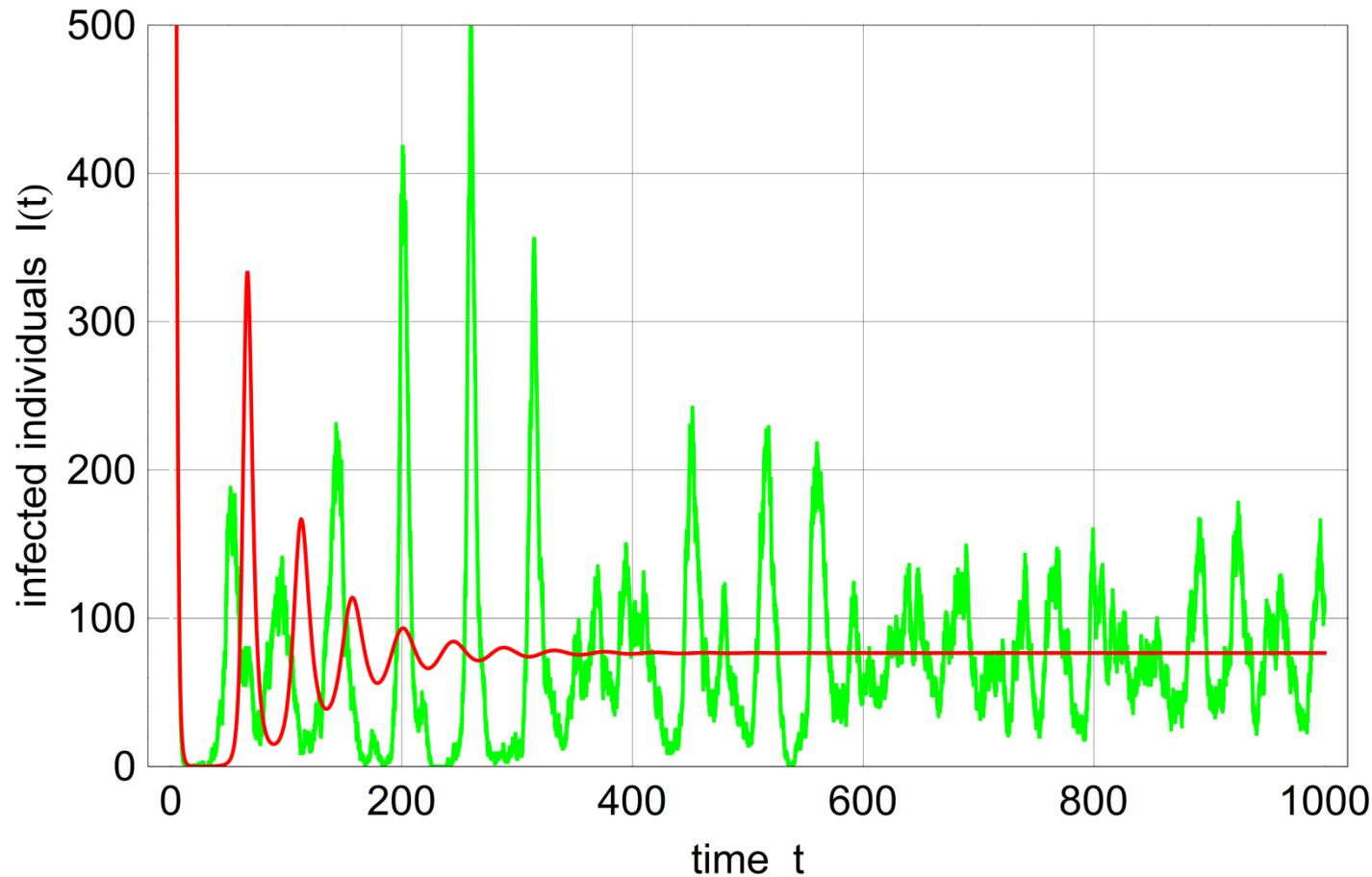
$$\frac{dE}{dt} = \beta S I - \varepsilon E - \mu E$$

$$\frac{dI}{dt} = \varepsilon E - \gamma I - \mu I$$

$$\frac{dR}{dt} = \gamma I - \mu R$$

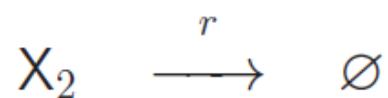
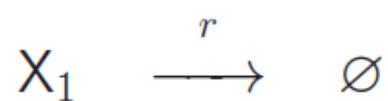
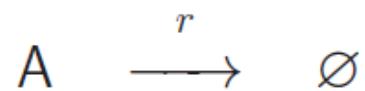
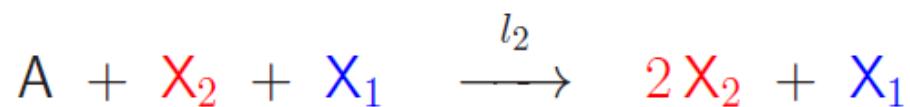
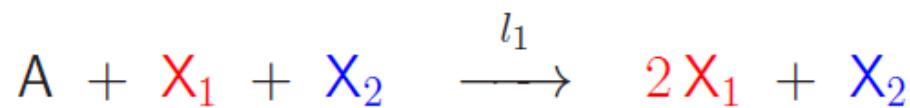
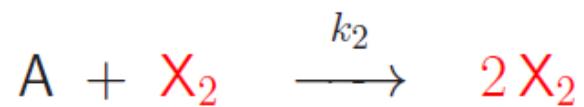
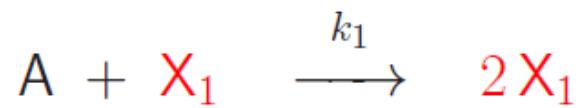
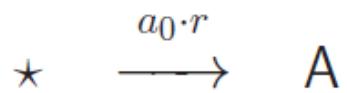
The SEIR model of theoretical epidemiology



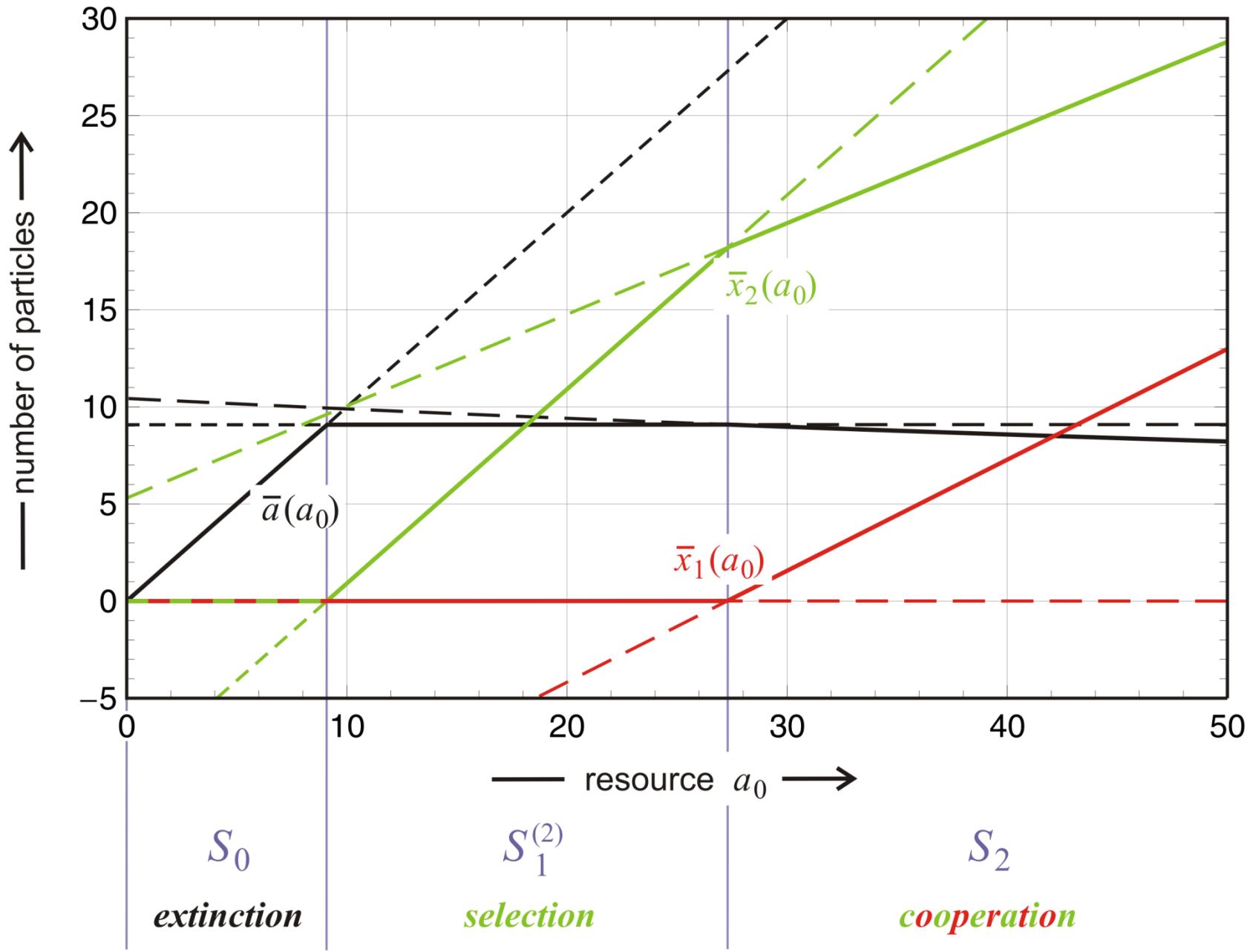


The SEIR model: **ODE integration** and **stochastic simulation**

Stochastic phenomena at small particle numbers

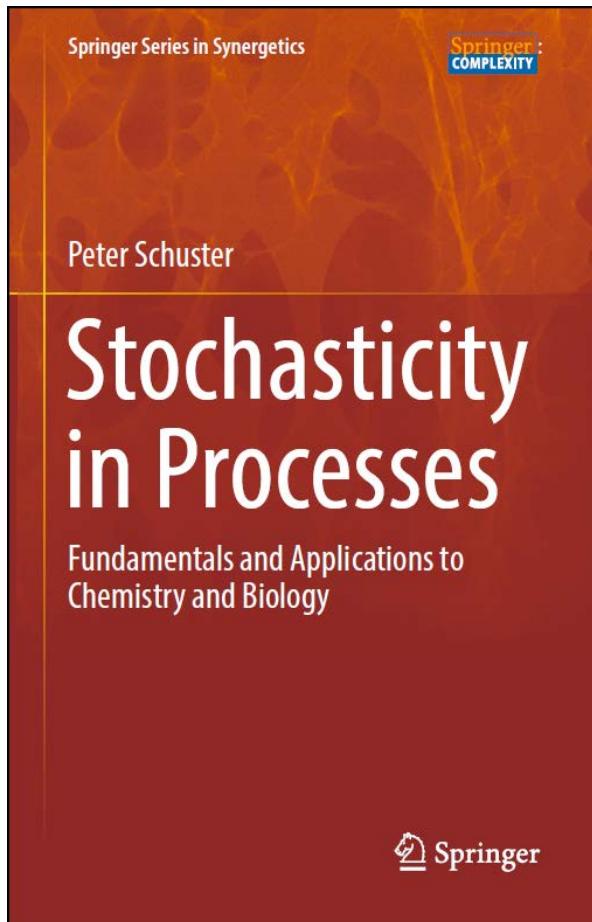


Reproduction and catalyzed reproduction of two species

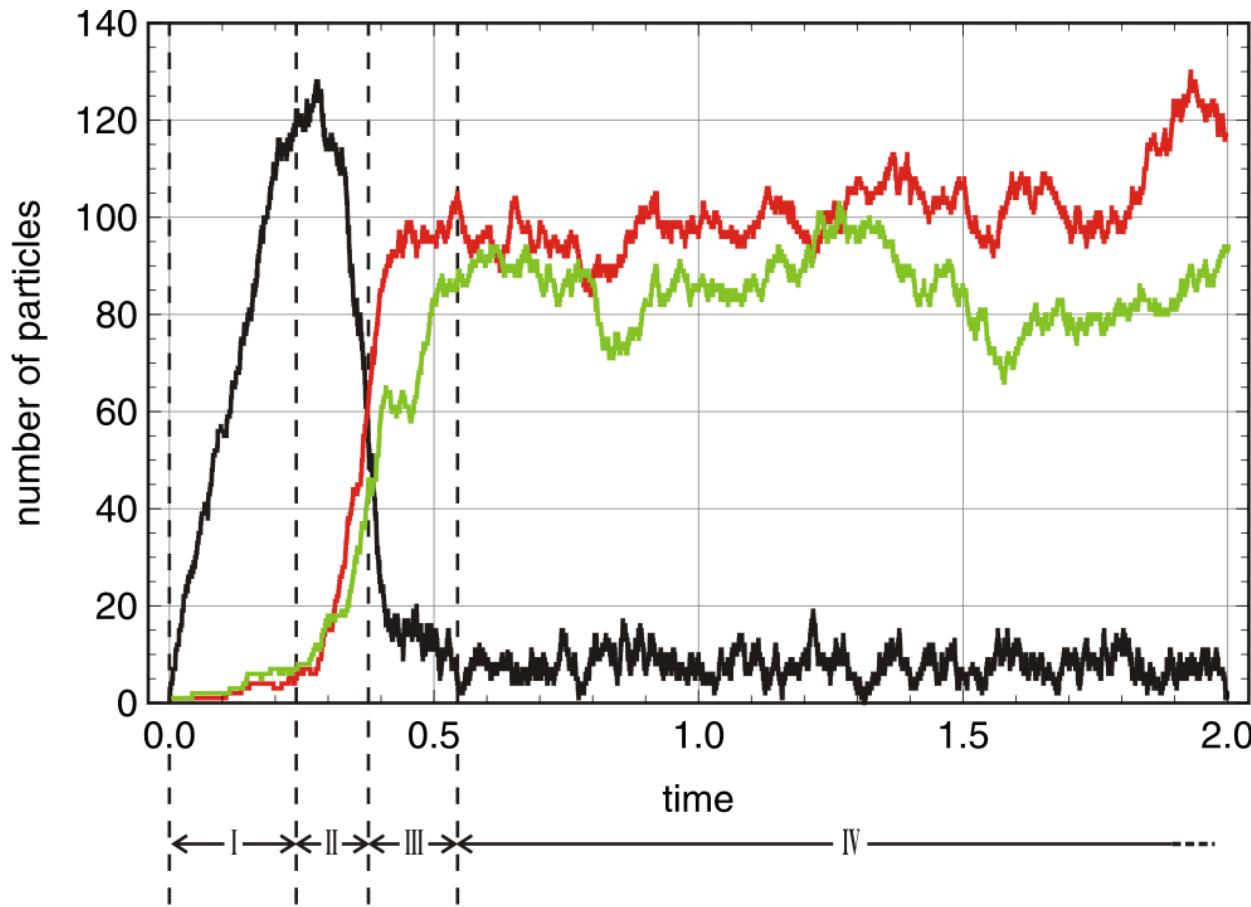


Stochastic modeling of chemical and biological systems

Stochastic simulation: D.T. Gillespie, Annu.Rev.Phys.Chem. 58:35-55, 2007



Peter Schuster. Stochasticity in Processes.
Fundamentals and Applications in Chemistry
and Biology. Springer-Verlag, Berlin 2016

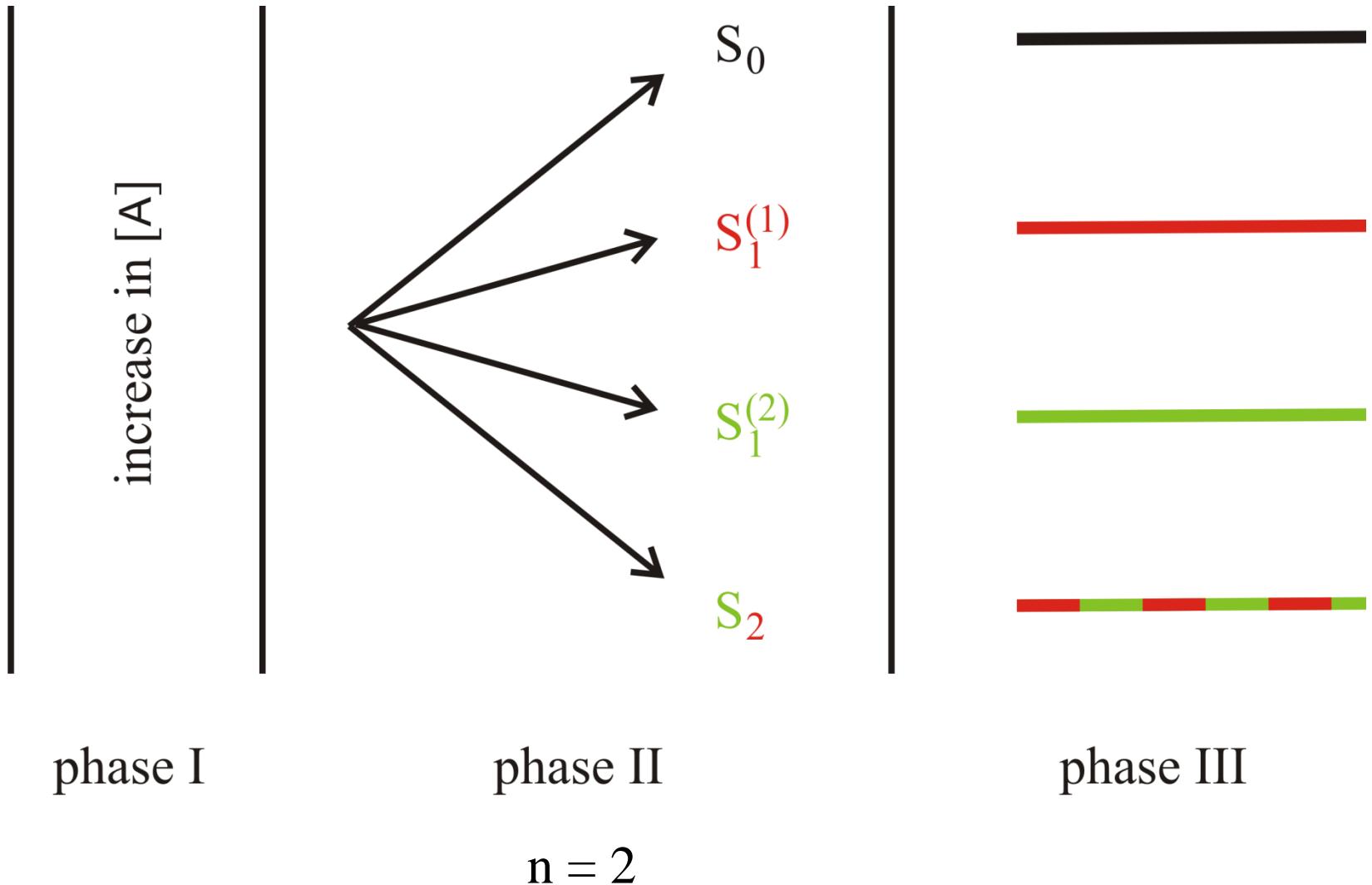


phase I: raise of [A]

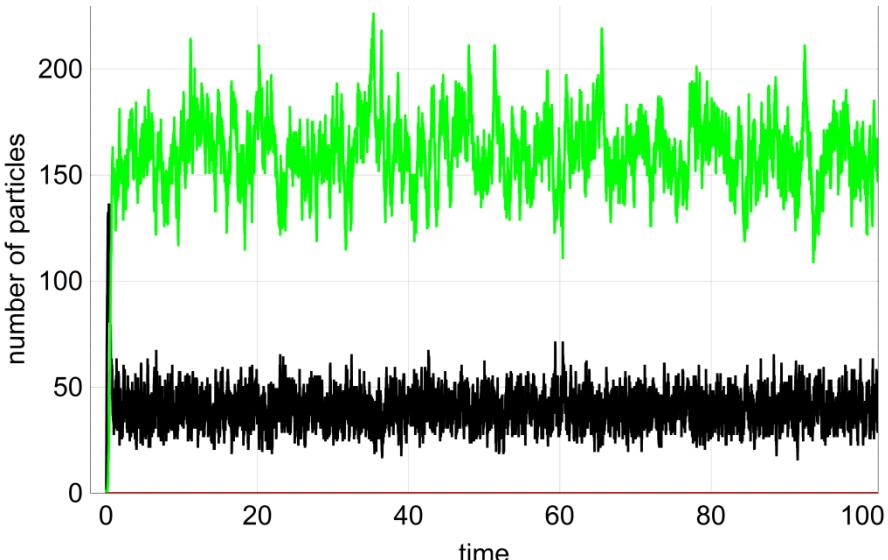
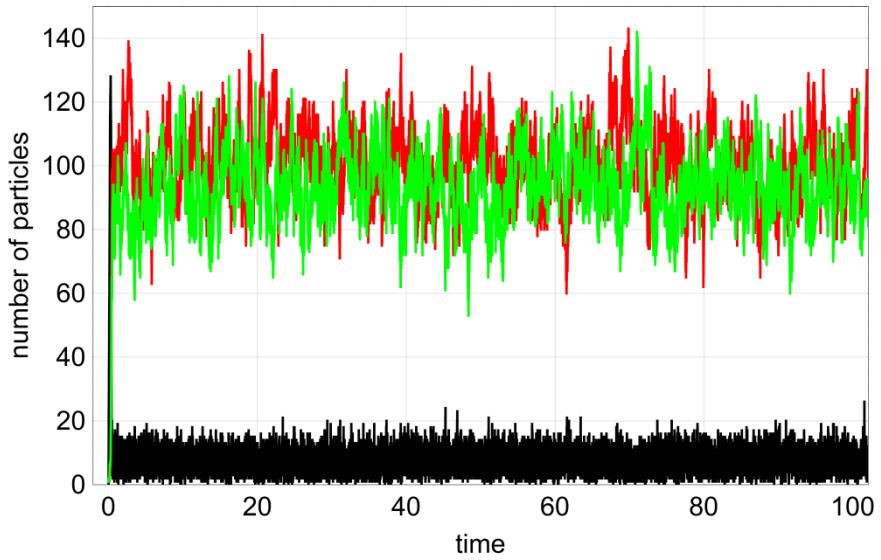
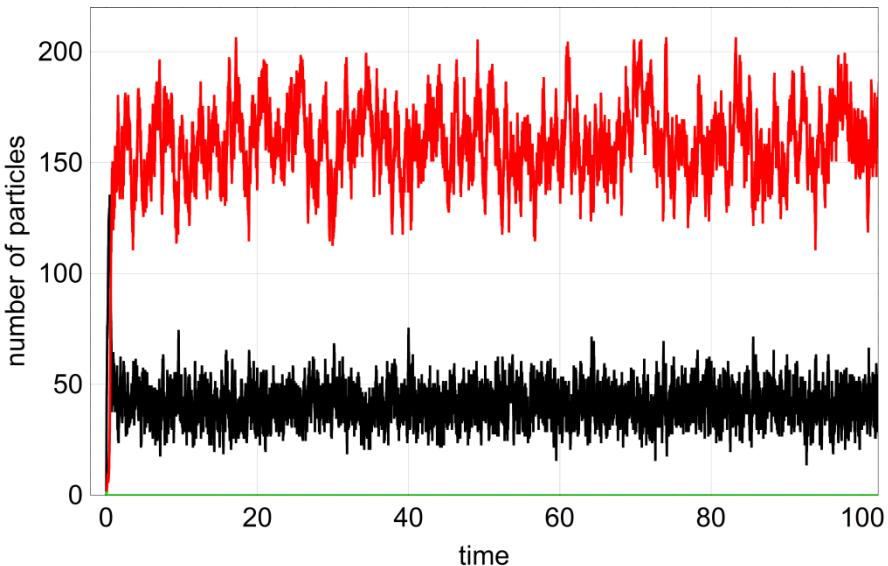
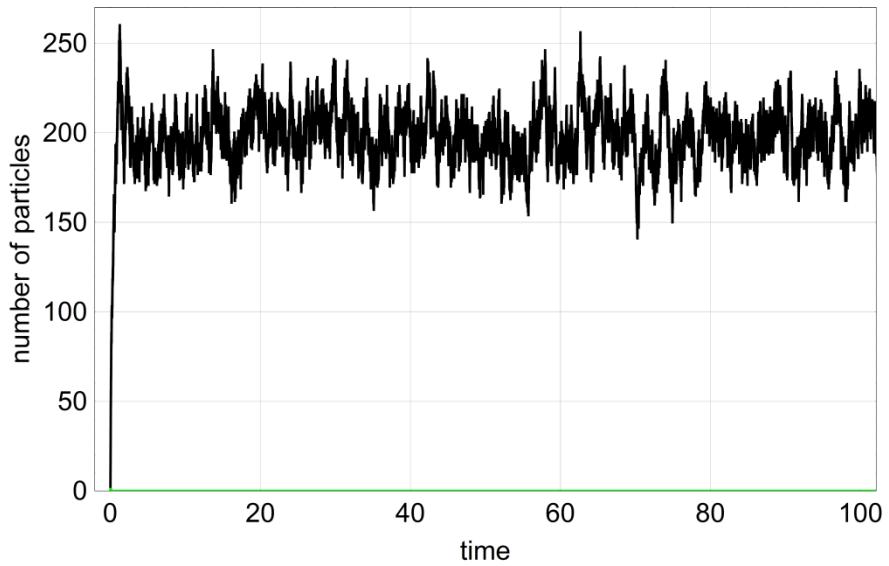
phase II: random choice of quasistationary state

phase III: convergence to the quasistationary state

phase IV: fluctuations around the quasistationary state



Random decision in the stochastic process



Competition and cooperation with $n = 2$

Initial values		Final states			
$X_1(0)$	$X_2(0)$	N_{S_0}	$N_{S_1^{(1)}}$	$N_{S_1^{(2)}}$	N_{S_2}
1	1	385.1 ± 23.6	1481.0 ± 36.8	1719.6 ± 37.8	6414.3 ± 53.8
2	2	14.9 ± 2.6	303.7 ± 16.0	354.5 ± 23.8	9326.8 ± 22.7
3	3	0	70.2 ± 10.0	106.2 ± 10.9	9823.4 ± 15.7
4	4	0	12.1 ± 2.6	28.0 ± 5.0	9959.9 ± 6.4

Choice of parameters: $k_1 = 0.011 \text{ [M}^{-1}\text{t}^{-1}\text{]}$; $k_2 = 0.009 \text{ [M}^{-1}\text{t}^{-1}\text{]}$;

$l_1 = 0.0050 \text{ [M}^{-2}\text{t}^{-1}\text{]}$; $l_2 = 0.0045 \text{ [M}^{-2}\text{t}^{-1}\text{]}$;

$a_0 = 200$; $r = 0.5 \text{ [Vt}^{-1}\text{]}$; $a(0) = 0$

Competition and cooperation with $n = 2$

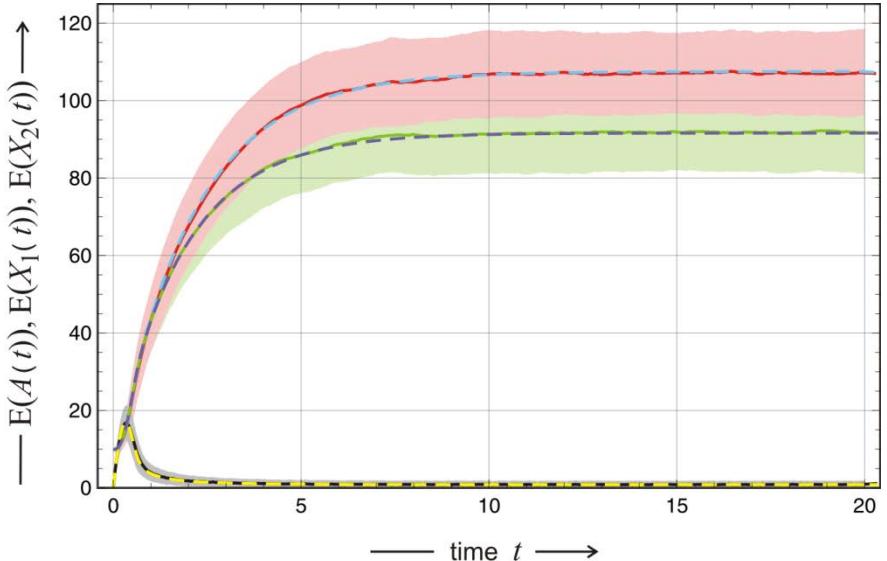
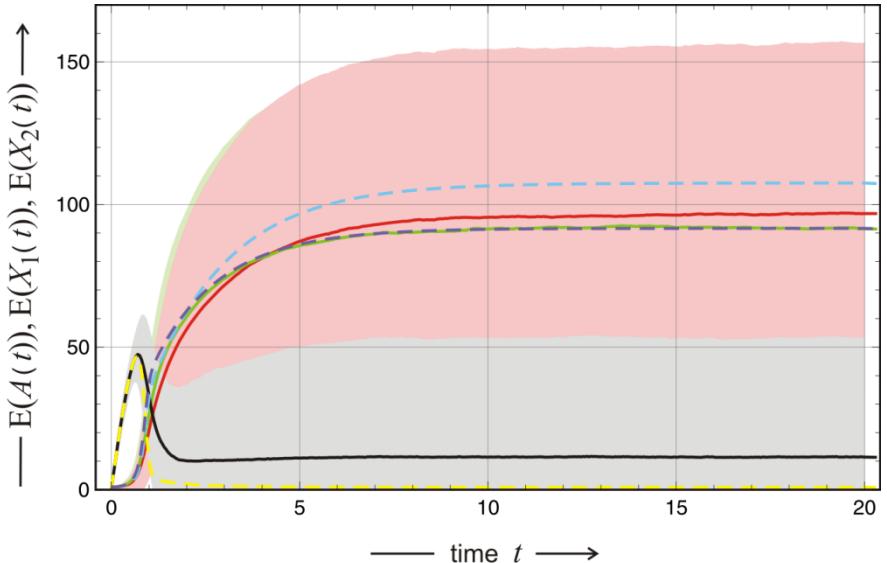
$$a(0) = 0, x_1(0) = x_2(0) = 1$$

expectation values and 1σ -bands

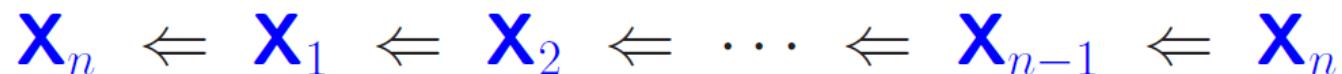
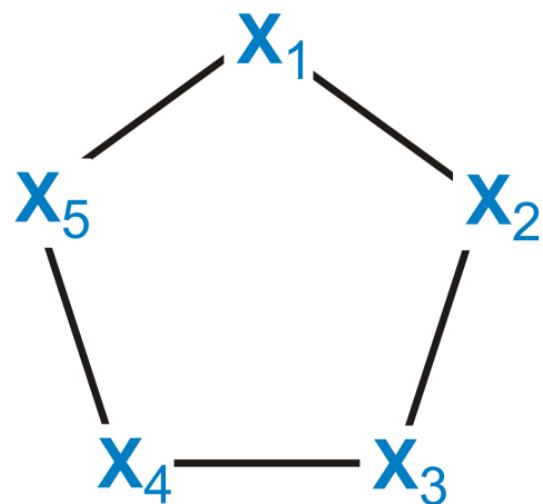
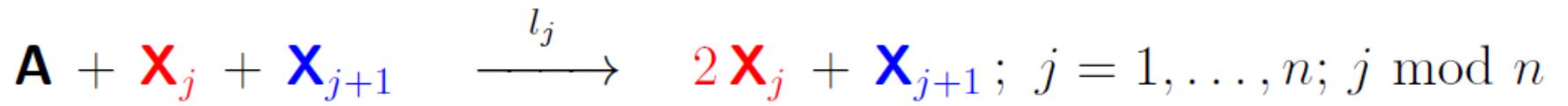
choice of parameters: $a_0 = 200$, $r = 0.5 \text{ [Vt}^{-1}\text{]}$

$$\begin{aligned} k_1 &= 0.09 \text{ [M}^{-1}\text{t}^{-1}], \quad k_2 = 0.11 \text{ [M}^{-1}\text{t}^{-1}], \\ l_1 &= 0.0050 \text{ [M}^{-2}\text{t}^{-1}], \quad l_2 = 0.0045 \text{ [M}^{-2}\text{t}^{-1}] \end{aligned}$$

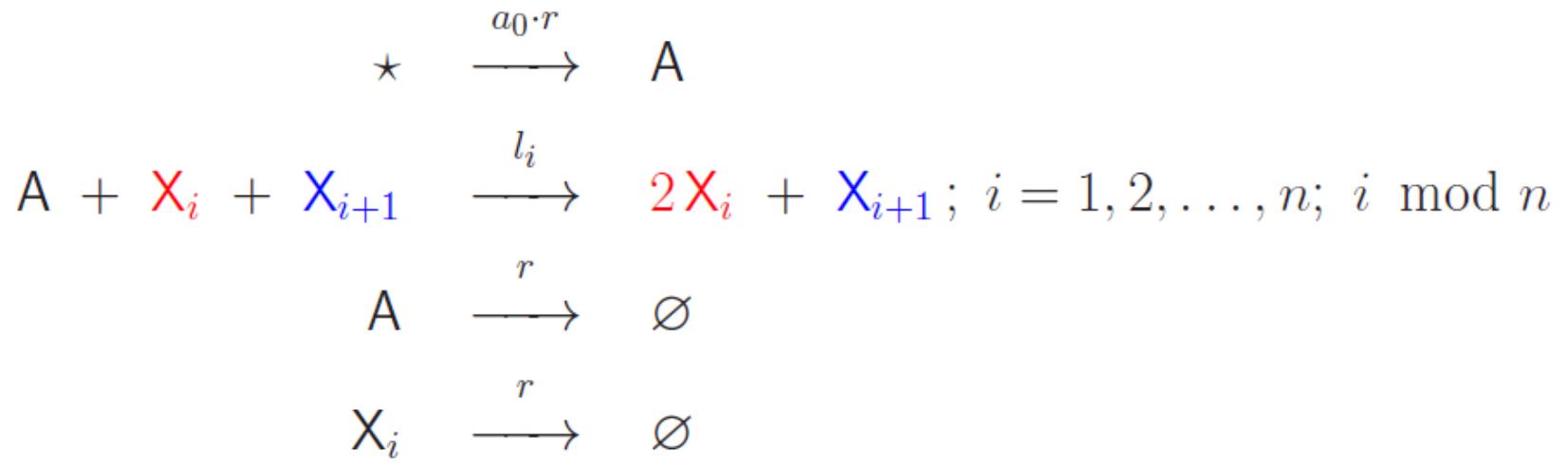
$$a(0) = 0, x_1(0) = x_2(0) = 10$$



Reintroduction of extinguished species through mutation

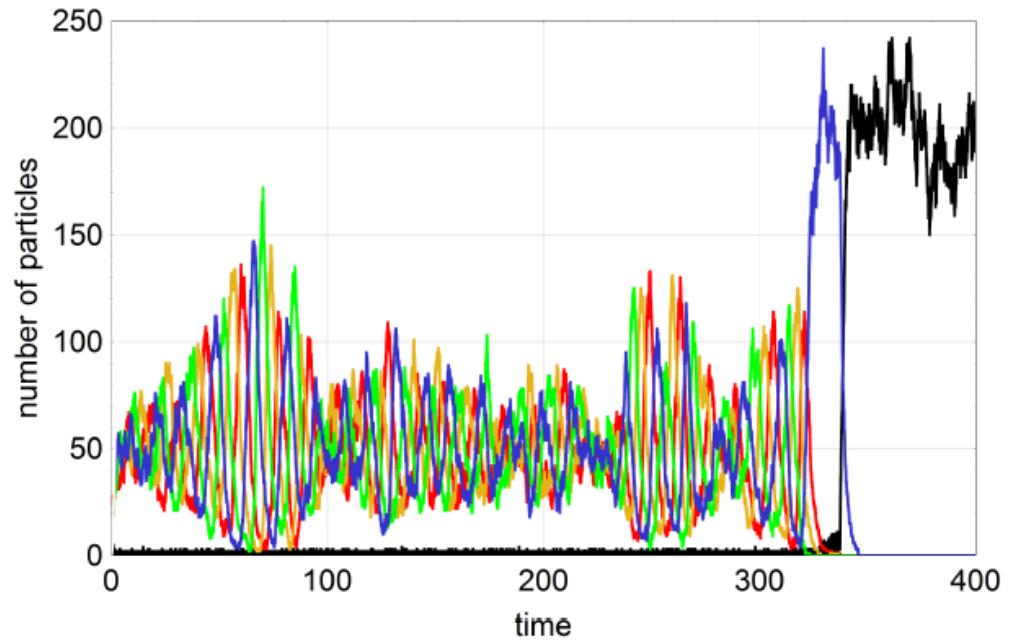


Catalytic hypercycles

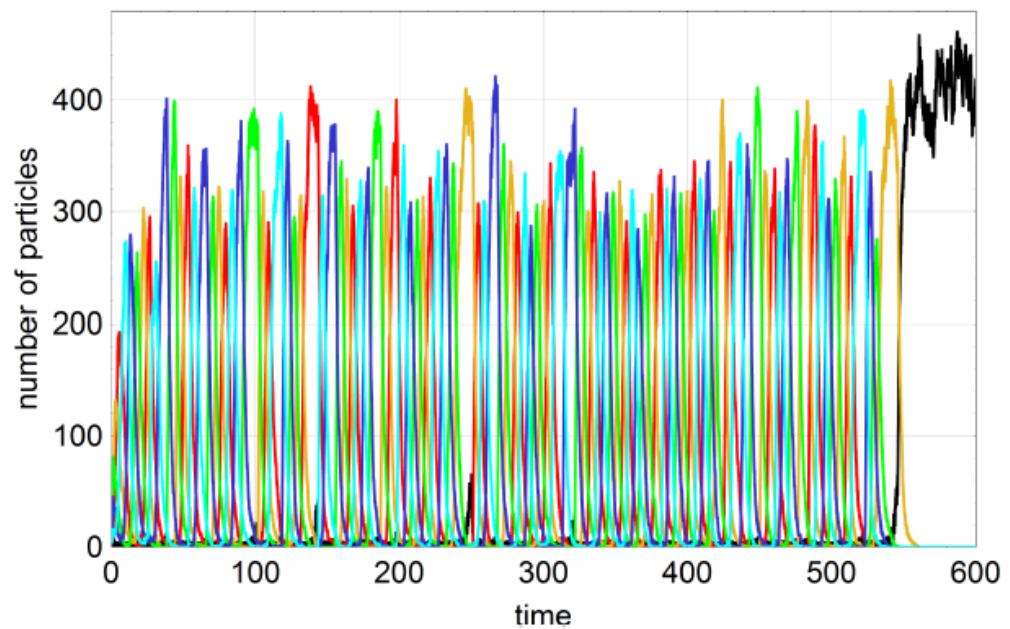


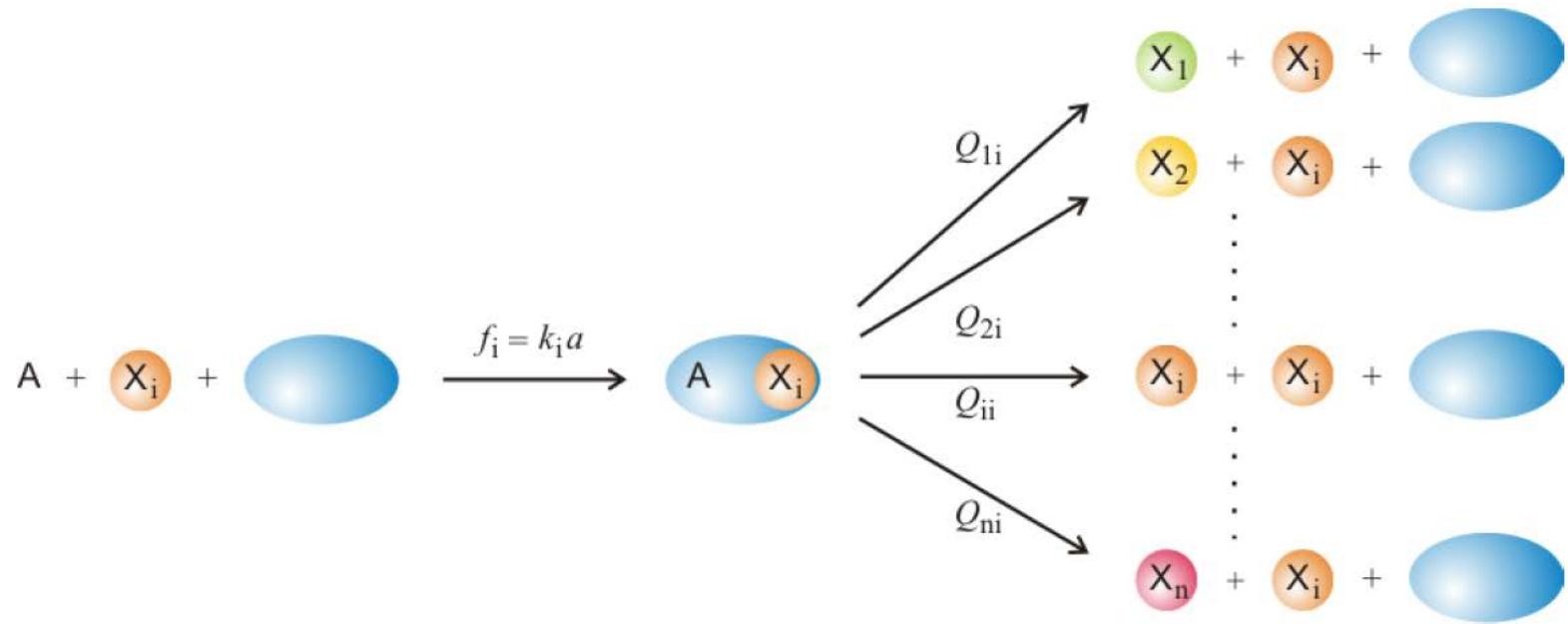
Second order autocatalysis: hypercycles in the flow reactor

$n = 4$

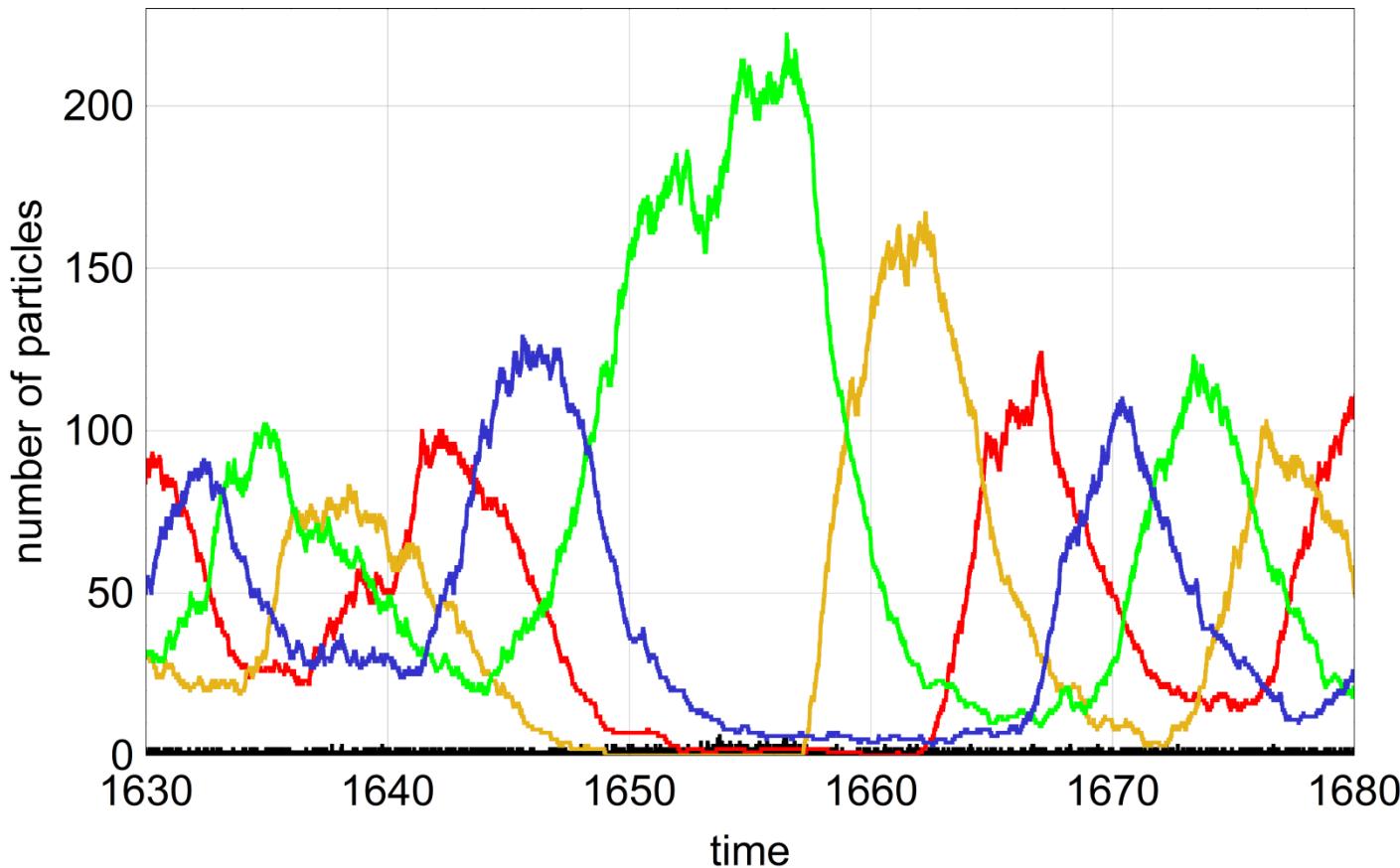


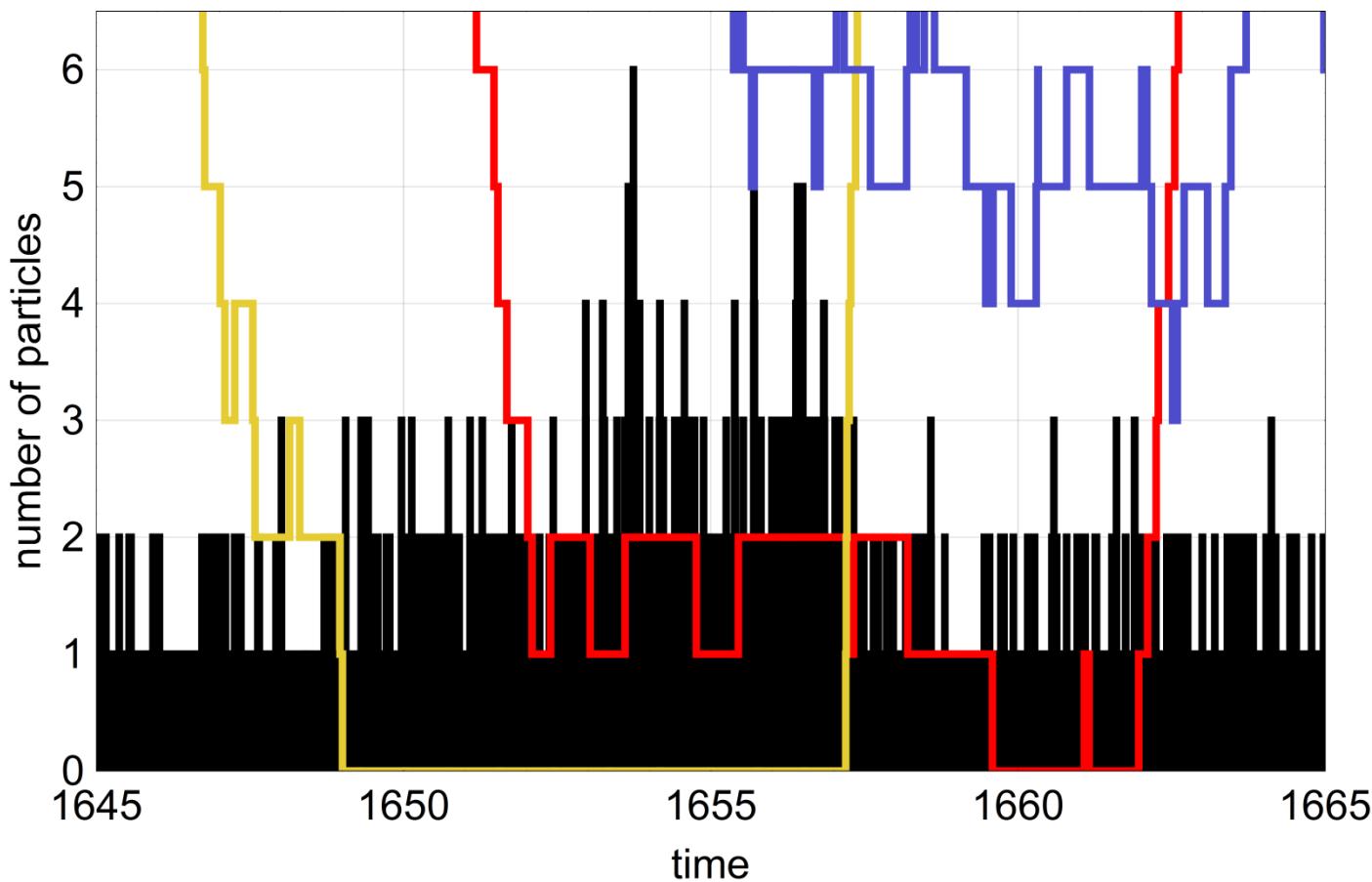
$n = 5$

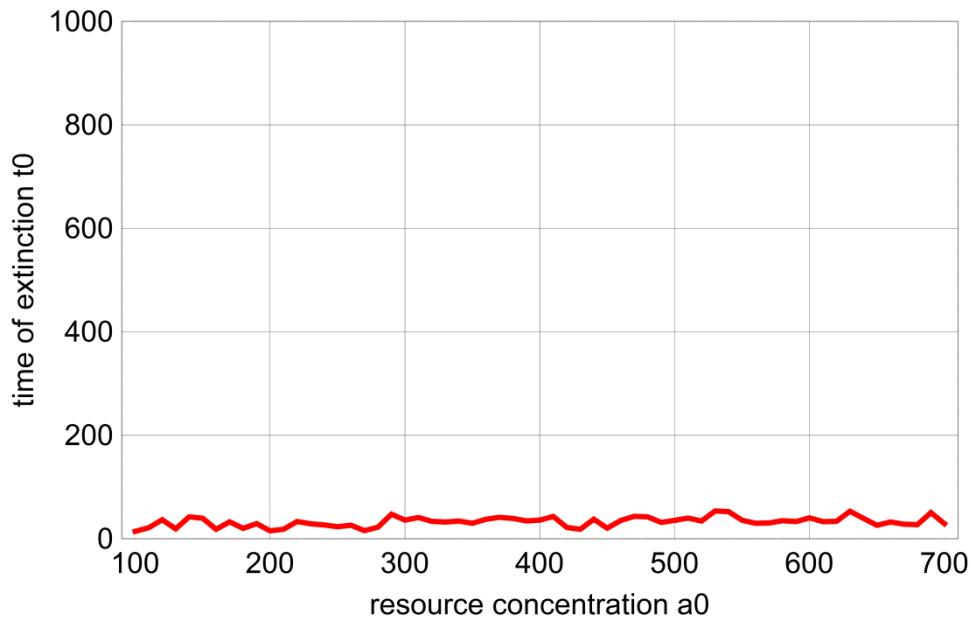




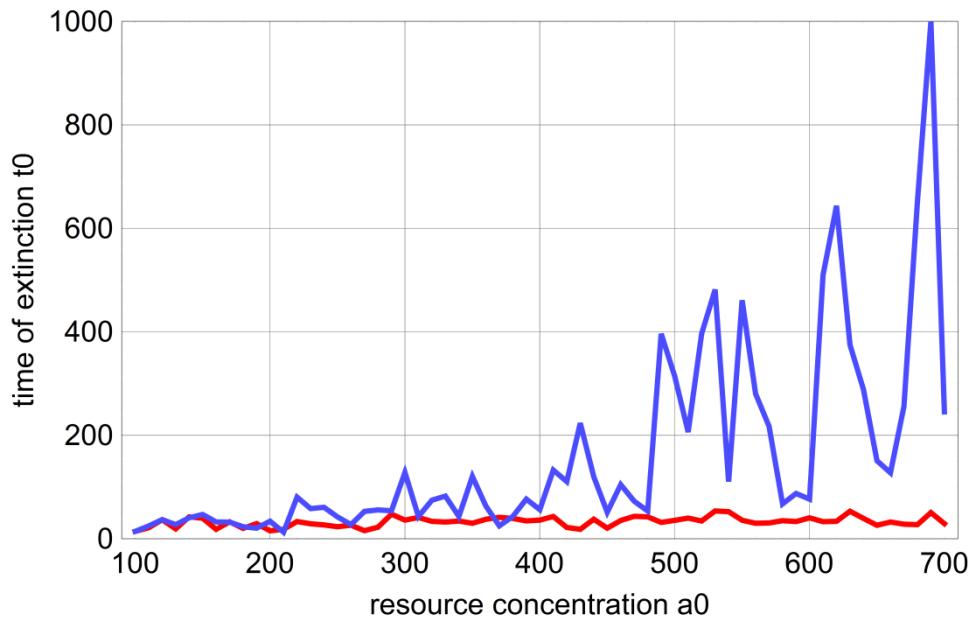
A molecular mechanism for mutation



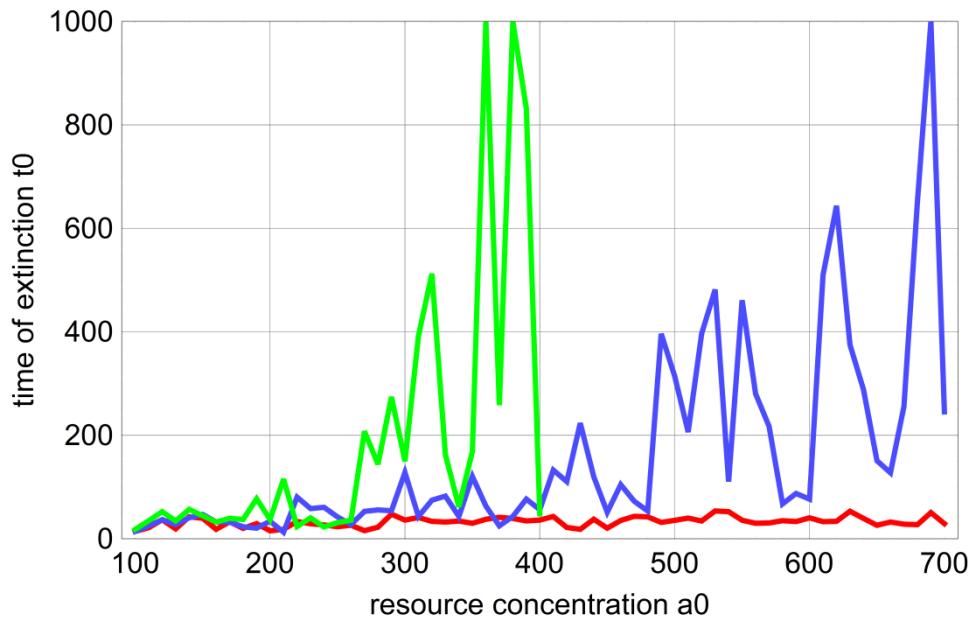




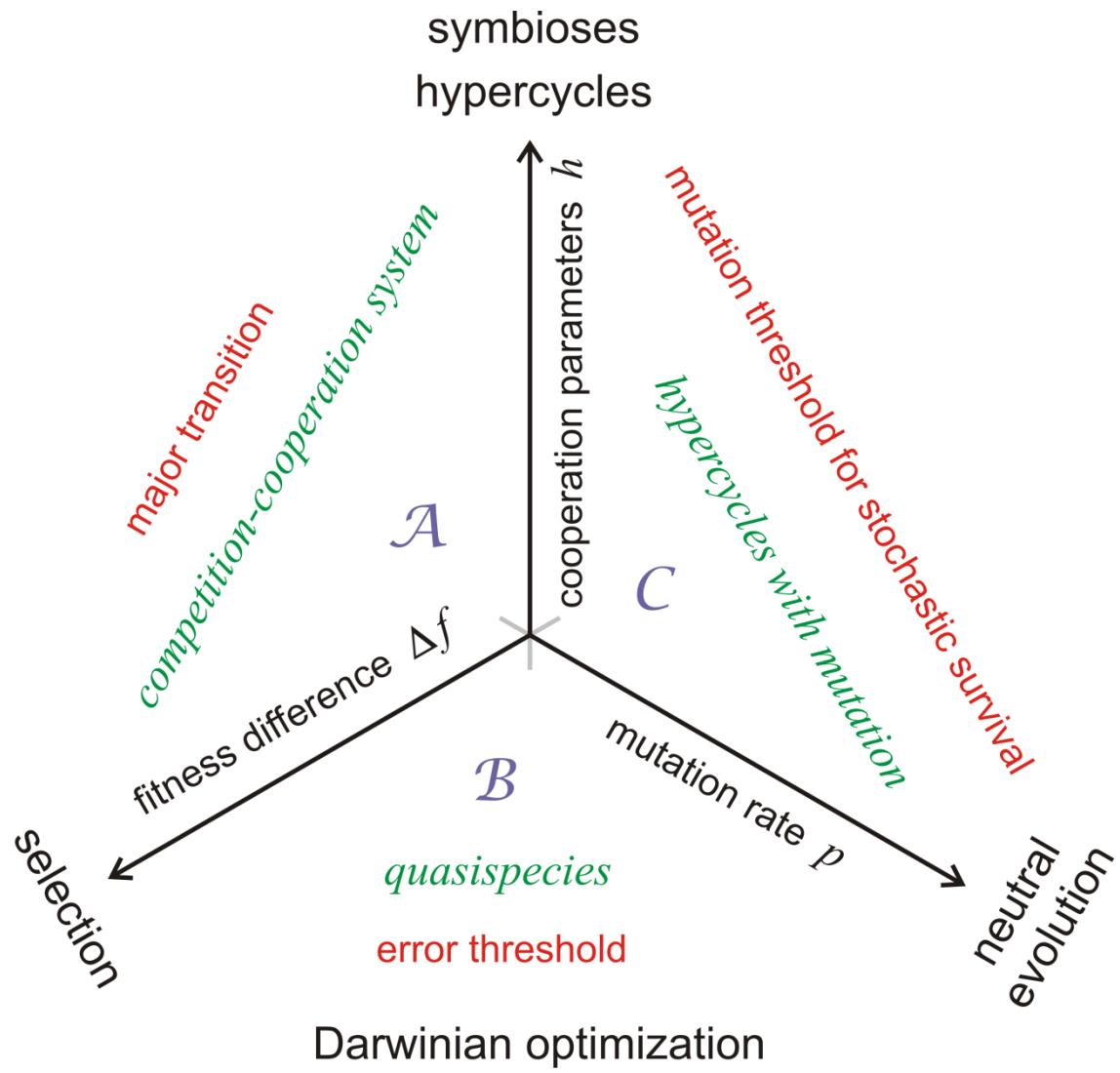
mutation rate: $p = 0.0000$



mutation rate: $p = 0.0010$



mutation rate: $p = 0.0020$



Danke für die Aufmerksamkeit!

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